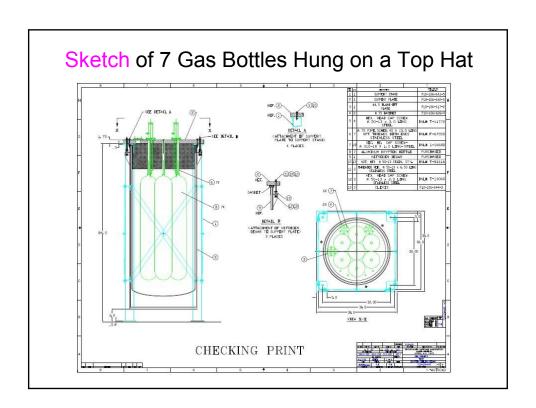
7 Left Over Bottles of Krypton Gas in Physics Dept.

K. C. Wu 9/30/10



Krypton, bottles and isolation valves

- Krypton is a rare gas
- Bottles CTC/DOT 3AL2015
- Isolation valves THERMO 2558, Cowin, L.A. Ca.
- A pressure gauge with 60 psi / 400 KPa is installed on top hat, but not connected to the bottles
- Each bottle has ~ 8 inch O.D. and ~ 48 inch height
- · Bottles have flat bottom

A chart on some gas cylinders

CYLINI SIZE	DOT SPECIFICATION	NOMINAL DIMENSIONS	TARE WEIGHT	INTERNAL VOLUME	REMARKS
49	3AA2400	24x140 cm. 9x55 in.	61 kgs. 135 lbs.	49.5 liters 3020 in. ³	Steel
44	3AA2265	23x130 cm. 9x51 in.	52 kgs. 115 lbs.	43.9 liters 2680 in. ³	Steel
-→ 30	3AL2015	20x122 cm. 8x48 in.	22 kgs. 48 lbs.	29.5 liters 1800 in. ³	Aluminum
16	3AA2015	18x84 cm. 7x33 in.	21 kgs. 47 lbs.	16.4 liters 1000 in. ³	Steel
8	3AA2015	18x46 cm. 7x18 in.	11 kgs. 24 lbs.	7.8 liters 475 in. ³	Steel
6	3AL2216	18x40 cm. 7x16 in.	7 kgs. 15 lbs.	5.9 liters 360 in. ³	Aluminum
3	3AA2015	10x43 cm. 4x17 in.	4 kgs. 9 lbs.	2.8 liters 172 in. ³	Steel
LB	3E1800	5x30 cm. 2x12 in.	0.75 kgs. 1.70 lbs.	0.46 liters 28.4 in. ³	Steel
SSLB	3E1800	5x30 cm. 2x12 in.	0.75 kgs. 1.7 lbs.	0.46 liters 28.4 in. ³	Stainless Steel
.150	3E1800	4x23 cm. 1.5x9 in.	0.6 kgs. 1.4 lbs.	0.15 liters 9.2 in. ³	Stainless Steel
.075	3E1800	4x13 cm. 1.5x5 in.	0.34 kgs. 0.75 lbs.	0.075 liters 4.6 in. ³	Stainless Steel

High-Pressure Aluminum Gas Cylinders High-Pressure Aluminum Gas Cylinders There are many different gas cylinder options available for packaging of Air Liquide specialty gas products. While most of our gas cylinders remain the property of Air Liquide, we also fill customer-owned cylinders provided they meet all appropriate safety requirements Outside **HP** Aluminum Approximate Capacity Water Volume†† DOT Cylinder Size inches lbs cu. in 2216 6909 9.8 33 15.6 16AL 3AL 2216 83 2350 7.25 30 958 15.7 7AL 3AL 2216 878 6.9 360 5.9 3AL PS 3AL 227 4.4 10.5 103 2015 3.5 1.7 8 1AL PS † For N₂ at 70°F 1 atm. † Nominal. Inst Resale cylinder only. 60"

Amount of Krypton in Physics Dept.

The 7 bottles of Krypton gas left from the 2001 experiment have been investigated. These bottles are DOT 3AL2015 Aluminum bottles. Each bottle has a nominal dimension of 8 inch OD and 48 inch height. Each bottle has an internal volume of 29.5 Liter.

From the P&I D and also confirmed by A. Hoffmann, each of these 7 bottles has an isolation valve on top and were connected together using a manifold above the top hat (no longer connected). Today, August uses a gas bottle regulator to measure pressures in each bottle. All show a pressure slightly above 1600 psi.

The volume at ambient temperature and pressure of Kr is approximately 138 times that at 1600 psi. Thus, the amount of Kr stored in the 7 bottles equals $\sim 7 \times 29.5 \times 138 \sim 28,000$ Liter.

As a side note, this Kr gas is equivalent to ~ 38 L of liquid Kr.

Price and "Condition" of Krypton – p1

1.2 / L for 7,500 L (~\$9,000) of Kr in a 44 L size bottle at ~ 130 bar -9/27/10

Per Pavel Perlov Global Director of Business Development Electronic Fluorocarbons, LLC tel. +1-508-435-7700 fax +1-508-625-1368 mobile: +1-617-592-3820

email: pavel@electronicfluorocarbons.com

\$1.6 / L for 1 \rightarrow 5,000 L and \$1.4 / L for > 5,000 L, \$1.3 / L > 10,000 L and \$1.25 / L > 20,000 L - 10/1/10

Scientific Grade?, Per Gerry Isenberg CTS Welco

Tel. 973-477-7145 Conversation with a representative of CTS Welco, (representing Praxair), scientific grade Kr has a shelf life of 60 months. Don't know the detailed, but one probably could assume Kr in the 7 bottles for \sim 20 years may not be 99.999% pure.

Price and "Condition" of Krypton – p2

Robert Pisani has the setup to transfer the gasses to another bottle. In addition he sent Mike a quote for the non research grade krypton that was \$7500 for a Size 1 cylinder (10k ltr) so, it has value. There is a possibility that Spectra gases may buy it from us in exchange for a credit. (\$ 0.75 / L?)

From BNL buyer J. Cafiero 10/1/2010

Yes I deal with Gerry for stock cylinders for a lot of other chemicals used here. Ok I received the pricing for various quantities which are listed below, so the cost per cylinder will vary based on the volume you are looking for. I can also have the rental and delivery fees waived, so the total PO cost would be for the Krypton only.

0 to 5000 Liters @\$ 1.60 5001 liters to 10,000 Liters @ \$1.40 10,001 liters to 20,000 liters @ \$1.30 > 20,001 liters and above @ \$1.25

Estimated Value of Krypton in Physics Dept.

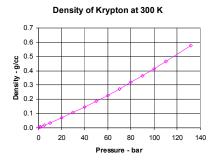
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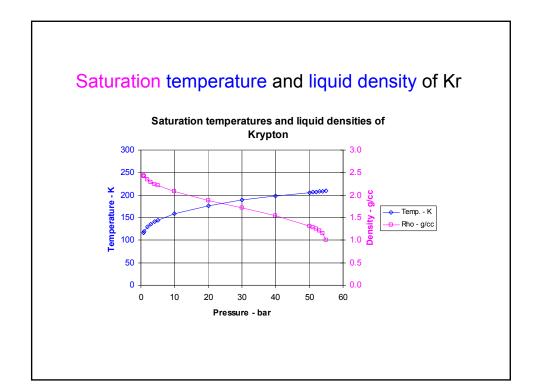
According to the business director of Electronic Fluorocarbons, the price equals $\sim \$1.2/L$ for 7500 L of 99.888% Kr in a 44 L bottle (~ 130 bar), or $\sim \$9,000$ to the bottle. For 99.99% Kr, the price is slightly lower (4 – 9%). These are the purchase price. In order words, we need $\sim \$33,000$ to buy 28,000 L of Kr today. As an independent check on price, the price Rob has for UHP (99.995%) grade is lower $\sim \$0.75$ / L from other vendor.

Density of Kr as a function of pressure at 300 K

P	Т	Р	rho	Р
bar	K	kPa	g/cc	psi
1	300	100	0.00337	14.5
2	300	200	0.00675	29.0
5	300	500	0.01697	72.6
10	300	1000	0.03430	145.1
20	300	2000	0.07004	290.2
30	300	3000	0.107	435.3
40	300	4000	0.146	580.4
50	300	5000	0.187	725.5
60	300	6000	0.229	870.6
70	300	7000	0.273	1016
80	300	8000	0.319	1161
90	300	9000	0.366	1306
100	300	10000	0.415	1451
110	300	11000	0.465	1596
131.6	300	13163	0.577	1910

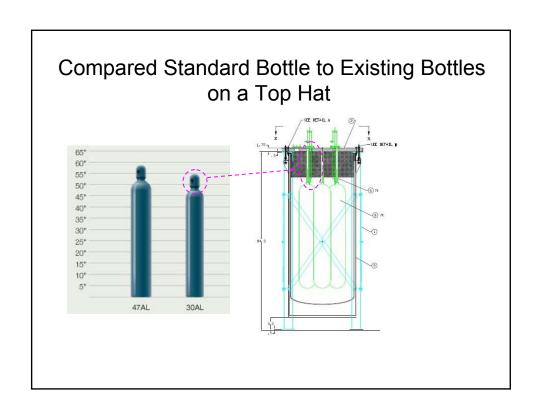


Note: 1 L of Kr → 3.37 gm



Bottles and Gas Transfer

 The following slides show modification made on existing gas bottle and mechanisms of transfer gas without a pump



CGA Valve Fittings

CGA Valve Fittings - Gases G through K

GAS	CGA Valve Outlet & Connection
"Genetron 21" (Dichlorofluoromethane)	660
"Genetron 23" (Fluoroform)	660
"Genetron 115" (Monochloropentafluoroethane)	660
"Genetron 152A" (1, 1-Difluoroethane)	510
"Genetron 1132A" (1, 1-Difluoroethylene)	350
Germane	350
Helium	580
Hexafluoroacetone	330
Hexafluoropropylene	660
Hydrogen	350
Hydrogen Bromide	330
Hydrogen Chloride	330
Hydrogen Fluoride	670
Hydrogen Selenide	350
Hydrogen Sulfide	330
lodine Pentafluoride	670
Isobutane	510
Isobutylene	510
Krypton	580

Gas Cylinder Safety - p1

Lessons Learned: Compressed Gas Cylinders Not Properly Stored

Effective Date: Sep 29, 2010 Point of Contact: Edward Sierra Identifier: 2010-OR-BWY12-0403 Provide Feedback

Lessons Learned Statement:

When compressed gas cylinders are not properly stored, it results in unsafe conditions that have the potential for injury to workers.

Discussion of Activities:

An Independent Assessment, Pressure Safety Program and Compressed Gas Cylinder Safety, was conducted at Y-12 National Security Complex/Production Facilities in September of 2009 to assess the effectiveness of implementation of the identified pressure safety and compressed gas cylinder safety procedures associated with the programs.

Emphasis was placed on the inspection and test of the pressure vessels and systems and the storage and handling of compressed gas cylinders.

The assessment identified several locations where compressed gas cylinders were not in compliance with Y73-400, Compressed Gas Cylinder Safety.

Gas Cylinder Safety – p2

Analysis:

The results of the Independent Assessment indentified the following:

- Cylinders stored without protective valve caps.
- Charged (Full) and empty compressed gas cylinders were not stored separately which made it cumbersome for delivery personnel to readily pickup and deliver cylinders in several locations.
- Cylinders were found missing the Compressed Gas Association label and/or other labeling identifying the gas within the cylinder in several different locations.
- 4. Legacy cylinders identified.
- 5. Storage issues identified:
 - Cylinders were inappropriately nested.
 - Cylinders were found improperly stored or secured.
 - · Cylinders not properly secured with chains.
 - Unknown and non-compatible gas cylinders were not segregated
 - Cylinders improperly secured or stored in the upright position.
- 6. Cylinders stored in front of electrical panels.
- Wooden skids of cylinders were stored on the ground where they were subject to being stuck by vehicular traffic.
- 8. At one location, storage of small compressed gas cylinders containing Oxygen and Acetylene as well as two portable Oxygen and Acetylene welding rigs were observed being stored together.

Transfer Kr gas from 7 x 29.5 L bottles to 5 x 43.9 L ones

res - psi	29.5 L BO)1) L			
roe nei		BO)2	BO)3	BO)4	BO)5	BO)6	BO)7	
ics - psi	1600	1600	1600	1600	1600	1600	1600	1600
0	643	1028	1258	1395				
	BO)1	BO)2	BO)3	BO)4	BO)5	BO)6	BO)7	
res - psi	643	1028	1258	1395	1600	1600	1600	
0	258	568	845	1066	1281	1409		
	BO)1	BO)2	BO)3	BO)4	BO)5	BO)6	BO)7	
res - psi	258	568	845	1066	1281	1409	1600	
0	104	290	513	735	955	1137	1323	
	BO)1	BO)2	BO)3	BO)4	BO)5	BO)6	BO)7	
res - psi	104	290	513	735	955	1137	1323	
0	42	142	291	470	665	854	1043	
	BO)1	BO)2	BO)3	BO)4	BO)5	BO)6	BO)7	
res - psi	42	142	291	470	665	854	1043	
0 .	17	67	157	283	436	604	781	1190
a	BO\1	BO)2	BU/3	BO)4	BO)5	BO)6	BO\7	
_	,	,	,	,	436	604	,	335
	o o o o o o o o o o o o o o o o o o o	0 643 BO)1 res - psi 643 0 258 BO)1 res - psi 258 0 104 res - psi 104 0 42 BO)1 res - psi 42 0 17 g BO)1	0 643 1028 BO)1 BO)2 res - psi 643 1028 0 258 568 BO)1 BO)2 res - psi 258 568 0 104 290 BO)1 BO)2 res - psi 104 290 0 42 142 BO)1 BO)2 res - psi 42 142 0 17 67 g BO)1 BO)2	BO)1 BO)2 BO)3 res - psi 643 1028 1258 BO)1 BO)2 BO)3 res - psi 643 1028 1258 BO)1 BO)2 BO)3 res - psi 258 568 845 0 104 290 513 BO)1 BO)2 BO)3 res - psi 104 290 513 0 42 142 291 res - psi 42 142 291 0 17 67 157 g BO)1 BO)2 BO)3	BO)1 BO)2 BO)3 BO)4 tres - psi 643 1028 1258 1395 BO)1 BO)2 BO)3 BO)4 tres - psi 643 1028 1258 1395 BO)1 BO)2 BO)3 BO)4 tres - psi 258 568 845 1066 BO)1 BO)2 BO)3 BO)4 tres - psi 104 290 513 735 BO)1 BO)2 BO)3 BO)4 tres - psi 104 290 513 735 BO)1 BO)2 BO)3 BO)4 tres - psi 42 142 291 470 BO)1 BO)2 BO)3 BO)4	BO)1 BO)2 BO)3 BO)4 BO)5 BO)1 BO)2 BO)3 BO)4 BO)5 Tres - psi 643 1028 1258 1395 1600 0 258 568 845 1066 1281 BO)1 BO)2 BO)3 BO)4 BO)5 Tres - psi 258 568 845 1066 1281 0 104 290 513 735 955 BO)1 BO)2 BO)3 BO)4 BO)5 Tres - psi 104 290 513 735 955 0 42 142 291 470 665 BO)1 BO)2 BO)3 BO)4 BO)5 Tres - psi 42 142 291 470 665 0 17 67 157 283 436 BO)1 BO)2 BO)3 BO)4 BO)5	BO)1 BO)2 BO)3 BO)4 BO)5 BO)6 BO)1 BO)2 BO)3 BO)4 BO)5 BO)6 Tres - psi 643 1028 1258 1395 1600 1600 0 258 568 845 1066 1281 1409 BO)1 BO)2 BO)3 BO)4 BO)5 BO)6 Tres - psi 258 568 845 1066 1281 1409 0 104 290 513 735 955 1137 BO)1 BO)2 BO)3 BO)4 BO)5 BO)6 Tres - psi 104 290 513 735 955 1137 0 42 142 291 470 665 854 Tres - psi 42 142 291 470 665 854 0 17 67 157 283 436 604 BO)1 BO)2 BO)3 BO)4 BO)5 BO)6	BO)1 BO)2 BO)3 BO)4 BO)5 BO)6 BO)7 res - psi 643 1028 1258 1395 BO)1 BO)2 BO)3 BO)4 BO)5 BO)6 BO)7 res - psi 643 1028 1258 1395 1600 1600 1600 BO)1 BO)2 BO)3 BO)4 BO)5 BO)6 BO)7 res - psi 258 568 845 1066 1281 1409 1600 0 104 290 513 735 955 1137 1323 BO)1 BO)2 BO)3 BO)4 BO)5 BO)6 BO)7 res - psi 104 290 513 735 955 1137 1323 res - psi 104 290 513 735 955 1137 1323 BO)1 BO)2 BO)3 BO)4 BO)5 BO)6 BO)7 res - psi 42 142 291 470 665 854 1043 res - psi 42 142 291 470 665 854 1043 res - psi 42 142 291 470 665 854 1043 g BO)1 BO)2 BO)3 BO)4 BO)5 BO)6 BO)7

Transfer Kr gas from 7 x 29.5 L bottles to 7 x 29.5 L ones

sring K	r gas from 7	7 old bottles	to 7 new b	ottles					
Old bot	tles	29.5 L	New bot	tles	29.5	i L			
Initially		BO)1	BO)2	BO)3	BO)4	BO)5	BO)6	BO)7	
	Pres - psi	1600	1600	1600	1600	1600	1600	1600	1600
BN)1	0	800	1200	1400	1500				
		BO)1	BO)2	BO)3	BO)4	BO)5	BO)6	BO)7	
	Pres - psi	800	1200	1400	1500	1600	1600	1600	
BN)2	0	400	800	1100	1300	1450			
		BO)1	BO)2	BO)3	BO)4	BO)5	BO)6	BO)7	
	Pres - psi	400	800	1100	1300	1450	1600	1600	
BN)3	0	200	500	800	1050	1250	1425		
		5014	5010	5010	5014	5015	5010	0017	
		BO)1	BO)2	BO)3	BO)4	BO)5	BO)6	BO)7	
DNI) 4	Pres - psi	200	500	800	1050	1250	1425	1600	
BN)4	0	100	300	550	800	1025	1225	1413	
		BO)1	BO)2	BO)3	BO)4	BO)5	BO)6	BO)7	
	Pres - psi	100	300	550	800	1025	1225	1413	
BN)5	0	50	175	363	581	803	1014	1213	
,-	Ü	00		000		000			
		BO)1	BO)2	BO)3	BO)4	BO)5	BO)6	BO)7	
	Pres - psi	5Ó	175	363	581	803	1014	1213	
BN)6	0	25	100	231	406	605	809	1011	
		BO)1	BO)2	BO)3	BO)4	BO)5	BO)6	BO)7	
	Pres - psi	25	100	231	406	605	809	1011	
BN)7	0	13	56	144	275	440	625	818	1261
Remain		BO)1	BO)2	BO)3	BO)4	BO)5	BO)6	BO)7	
	Pres - psi	13	56	144	275	440	625	818	339

Other information

• The following slides are for references.

Physical Constants of Kr

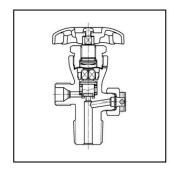
PHYSICAL CON	ISTANTS
Chemical name	Kr
Molecular weight	83.80
Density of the gas at 70°F (21,1°C), 1 atm	0.2172 lb/ft3, 3.479 kg/m3
Specific gravity of the gas at 70°F (21,1°C), 1 atm	2.899
Specific volume of the gas at 70°F (21,1°C), 1 atm	4.604 ft3/lb, 0.287 m3/kg
Boiling point at 1 atm	-244.0°F, -153.4°C
Melting point at 1 atm	–251°F, −157°C
Critical temperature at 1 atm	-82.8°F, -63.8°C
Critical pressure	798.0 psia, 55.02 bar
Critical density	56.7 lb/ft3, 908 kg/m3
Triple point	-251.3°F, -157.4°C
Latent heat of vaporization at normal boiling point	46.2 Btu/lb, 107.5 kJ/kg
Latent heat of fusion at triple point	8.41 Btu/lb, 19.57 kJ/kg
Specific heat of the gas at 70∘F (21,1∘C), 1 atm Cp	0.060 Btu/(lb) (°F) 0.251 kJ/(kg) (°C)
Cv	0.035 Btu/(lb) (°F) 0.146 kJ/(kg) (°C)

Shipping Data - Kr

	SHIPPING DATA					
Synonyms	Kr					
CAS Register Number	7439-90-9					
DOT Classification	Nonflammable gas					
DOT Label	Nonflammable gas					
Transport Canada Classification	2.2					
Substance Identification (SI)	1056					
UN Number	UN 1056					
Hazards	High Pressure and suffocation					
Toxicity (TLV)	Asphyxiant					
Flammability Range (in air)	Nonflammable gas					
Odor	None					

New Thermo Valve

- Available with a wide range of CGA Outlets
 Available with a cap type safety assembly having fuse metal backed or non-backed copper rupture discs
 In full compliance with CGA and DOT regulations.
- regulations



- Seat material: 15% glass filled Kel-F (Neoflon, Daiflon)
 Optional seat material: Pure Kel-F (Neoflon, Daiflon)

Nylon Vespel

Vespel

• Below usually held in stock at Thermo for rapid shipment:

CGA 580 with 3775 psi unbacked safety assembly

CGA 580 with 4000 psi unbacked safety assembly

CGA 590 with 3775 psi unbacked safety assembly

CGA 350 with 3775 psi 165°F fusible backed safety assembly

CGA 350 with 4000 psi 165°F fusible backed safety assembly

RIX oil-free compressor to ~ 2200 psi

Microboost High Pressure Oxygen Compressors Check up to five results to perform an action.

- Maximum Horsepower. 0.5 HP
 Max Discharge Pressure: 2,200 PSIG
 Flow Ranges: 46 16 SCFH
 Cooling: AIR
 GAS: 02, He, N2, OTHER DRY GASES

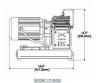
RIX Oxygen Compressors have been developed specifically for the exacting requirements of safe, reliable oxygen compression service. The Microboost Series incorporates design features developed over several decades for a variety of industrial and military oxygen compressor applications. These compressors are well suitled for the compression of oxygen, helium nitrogen, and other clean, dry gases to 2200 psig. The RIX Microboost high-pressure oxygen compressor will fill up to 25 "D" cylinders (15 cu fl / 425 Liters) in a 24-hour period.

The Microboost is a three-stage reciprocaling compressor employing a stepped, oil-free piston. The crankshaft is driven via a speed reducing gearbox on a vertical design crankcase. Heat exchangers, crankcase and compression cylinders are air-cooled.

- High pressure for Oxygen, Helium, Nitrogen and other dry gases including cylinder fill applications.
 Discharge Pressure to 2200 PSIG (152 Barg)

Take a Look at Our Microboost Brochure





Gas Cylinder Comparison Chart

Gas Cylinder Comparison Chart

Air Liquide	Scott	Airgas	Linde	Matheson	Praxair
47AL	KAL	2	2		AT
47AL 30AL	AL	150A	A31	1R	AS
16AL	BL	80A	A16	2R	AQ
7AL	CL	33A	A07	3R	AG
3AL	-	-	-		A3
1AL	-	1-1	100	1 - 1	-

Cylinder Comparison Chart - Airgas

Appendix

Airgas.

Specialty Gas Cylinder Size Comparison Chart

Approximate Dimensions (inches)	Airgas	Linde	Air Liquide	Praxair	Matheson Trigas	MG	Scott Specialty Gases
High Pressure Steel							
9 x 55	300	049	49	Т	1L	300	K
9 x 51	200	044	44	K	1A	200	A
7 x 33	80	016	16	Q	2	80	В
7 x 19	35	007	7	G	3	35	С
2 x 12	LB	LBR	LB	LB	LB	LB	LB
4 x 26	E	005	MEDE	ANE	3L	E	ER
10 x 51	3HP	485	44H	6K	10	3HP	19-8
9 x 51	2HP	-	44H	зк	1H	2HP	
Aluminum							
10 x 52	300A	_	AT		_		_
8 x 48	150A	A31	30AL	AS	1R	150AL	AL
7 x 33	80A	A16	22AL	AQ	2R	AAL	BL
7 x 16	33A	A07	7AL	AG	3R	33AL	CL
4.5 x 15	9A	-	9AL	-	-	9AL	

Cylinder Comparison Chart – Matheson Tri-Gas

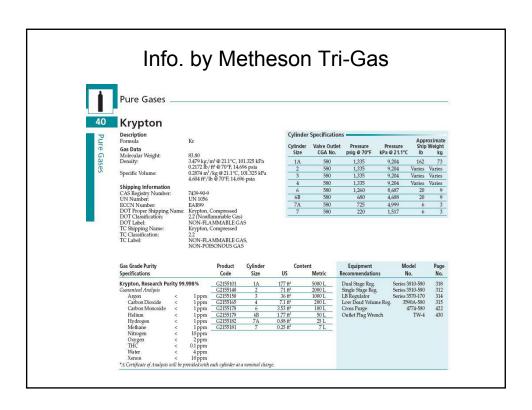
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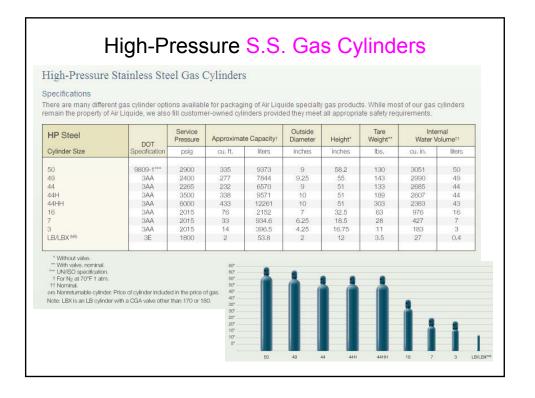
Cylinder Information

Cylinder Comparison Chart

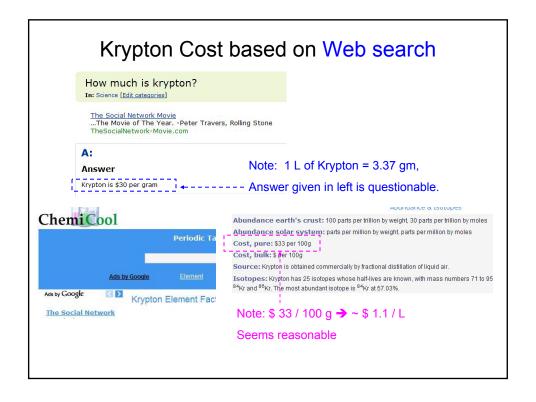
	son Trl-Gas	Nominal	Material of						
Specialty	Electronics	Dimensions	Construction	AGT	Airgas	Air Liquide	Linde	Air Products	Praxair
B1		30x53	S	1/2Ton				A-5	HT
1F		15x50	S	LP30	350	110	110	A-1	PX/FX
		12x43	S	LP15	65	55		A-3	FC
HF		12x18	S	LP05	25	22LP			
		9x36	S	LP05					
		8x9	S	LP01					
1L	QK	9x55	S	49	300	49	049	A	T
1A	QA	9x51	S	44	200	44	044	В	K
1R	QX	8x48	A	29AL	150A	30AL		B(AL)	AS
2	GA	9x26	S	16	80	16	016	C	Q
2R	GX	7x33	A	16AL	80A	22AL		C(AL)	AQ
3	UA	6x19	S	7	35	7	007	D-1	G
3R	UX	7x16	A	6AL	33A	7AL		D-1(AL)	AG
4	JA	4x13	S	3	10	3	003	D	F
LB		2x12	S	LB		LB		LB	LB

A= Aluminum; S= Steel









Krypton Gas

Pure Gas: Krypton

DESCRIPTION

Krypton is a rare atmospheric gas which is odorless, colorless, tasteless, nontoxic, monatomic and chemically inert. The concentration of Krypton in the atmosphere by volume percent is 1.1 x 10-4. Krypton is principally shipped and used in gaseous form for excimer lasers, light bulbs, window insulation and R & D laboratory research. Spectra Gases Material Safety Data Sheets (MSDS) are available for Krypton gas and should be used as guidelines in regard to first aid, methods of storage, handling and general use of Krypton.

PURITY SPECIFICATIONS (MAXIMUM IMPURITY LEVELS)*						
Contaminant	Research Grade 99.999%	UHP Grade 99.995%				
Argon (Ar)	2.0 ppm	3.0 ppm				
Carbon Dioxide (CO ₂)	0.5 ppm	1.0 ppm				
Carbon Tetrafluoride (CF ₄)	0.5 ppm	1.0 ppm				
Hydrogen (H₂)	0.5 ppm	3.0 ppm				
Krypton (Kr)	0.5 ppm	1.0 ppm				
Neon (Ne)	0.5 ppm	3.0 ppm				
Nitrogen (N ₂)	2.0 ppm	3.0 ppm				
Oxygen (O ₂)	0.1 ppm	1.0 ppm				
Water (H ₂ 0)	0.2ppm	1.0 ppm				
Xenon	5.0 ppm	25.0 ppm				

^{*} Higher purities are available upon request.

Info. By Spectra Gases

	CYLINDER INFORMATION								
Purity	Cylinder Size*	Valve Outlet*	Volume Liters	Gross Weight Lbs/Kg	Pressure Psig/Bar				
	1	580	10000	213 / 97	2300 / 160				
	2	580	5000	155 / 70	1350 / 94				
Decemb Condo	3	580	2000	63 / 29	1500 / 104				
Research Grade	4	580	1000	31 / 14	1450 / 101				
	5	580	500	16/ 7	2000 / 139				
	LB	580/170	50	6/3	1400 / 98				
	1	580	10000	213 / 97	2300 / 160				
	2	580	5000	155 / 70	1350 / 94				
IIIID C I	3	580	2000	63 / 29	1500 / 104				
UHP Grade	4	580	1000	31 / 14	1450 / 101				
	5	580	500	16/ 7	2000 / 139				
	LB	580/170	50	6/3	1400 / 98				
	D1	580	400	16/7	1300 / 91				
	D2	580	200	11/5	1050 / 73				
	D2	580	100	10/5	575 / 41				
Non-Refillable Cylinders	D3	580	50	7/3	675 / 48				
*	D3	580	25	6/3	350 / 25				
	D7	580	20	3/1	240 / 18				
	D7	580	12	3/1	140 / 11				

^{*} Additional cylinder sized and/or valve outlets are available upon request.

(Continued)



Branchburg, New Jersey 08876 USA. tel: 1.908.252.9300. toll free: (US & Canada) 1.800.932.0624. www.spectragases.com

4



Thermo Valve (new type?)

G55 Brass Diaphragm Valves For Specialty Gases



- Forged Brass body and brass internal components for high purity gases
 Five (5) Phosphor Bronze diaphragms for high vacuum and high pressure integrity
 Designed for easy open and easy close operation

Info. on some common gas

Spec vol at 70°F and 1 atm ► cylinder designation ▼	NH ₃ 22.5 ft ³ /lb		Ar 9.7 ft ³ /lb		He 96.7 ft ³ /lb		8.74 ft ³ /lb		H ₂ 192.0 ft ³ /lb		N ₂ 13.8 ft ³ /lb		O ₂ 12.1 ft ³ /lb		Air 13.3 ft ³ /lb	
	AA	15 x 52	114													
A	10 x 49	114	9 x 55	2640	9 x 55	2640			9 x 55	2640	9 x 55	2640	9 x 55	2640	9 x 55	2640
A(AL)			10 x 52	2000							10 x 52	2200				
В	9 x 51	114	9 x 51	2490	9 x 51	2490	9 x 51	830	9 x 51	2265	9 x 51	2492				
B (AL)	8 x 48	114	8 x 48	2000			8 x 48	830	8 x 48	2000	8 x 48	2000			8 x 48	2000
BX			10 x 51	6000	10 x 51	6000			10 x 51	6000	10 x 51	6000				
BY			9 x 51	3500	9 x 51	3500			9 x 51	3500	9 x 51	3500				
С	8 x 22	114	7 x 33	2215	7 × 33	2215			7 × 33	2015	7 x 33	2015	7 x 33	2215	7 × 33	2215
D			4 x 17	2215	4 x 17	2215	4 x 17	830	4 x 17	2015	4 x 17	2015	4 x 17	2215		
D-1	7x 19	114	7 x 16	2000					7 x 19	2015						
D-1 (AL)	7 x 16	114														
D	4 x 17	114														
L.B.			2 x 12	1,800	2 x 12	1800					2 x 12	1800	2 x 12	1800		
L.B.I.	2 x 12	114														
4X(AL)-100			4 x 10	850	4 x 10	900			4 x 10	2000	4 x 10	850	4 x 10	850		
4X(AL)-50			4 x 10	425							4 x 10	425				

Info. on some common gas

gas	NH ₃	Ar	He	CO ₂	H ₂	N ₂	02	Air	
color/odor	none/strong	none/none	none/none	none/none	none/none	none/none	none/none	none/none	
flamm	mod	not	not	not	YES 4 75%	not		**	
Toxic	yes	asphyxiant	asphyxiant	asphyxiant	asphyxiant	asphyxiant	*		
TWA	25 ppm (ACGIH)	none		5,000 ppm (ACGIH)					
MW	17.03	39.95	4.003	44.01	2.016	28.01	32.00		
DOT Haz Class	Nonfam Gas (2.2)	Nonflam Gas (2.2)	Nonfam Gas (2.2)	Nonfam Gas (2.2)	Flam Gas (2.1)	Nonfam Gas (2.2)	Nonflam Gas (2.2)	Nonflam Gas (2.2)	
DOT Label	Nonflammable Gas	Nonflammable Gas	Nonfammable Gas	Nonflammable Gas	Flammable Gas	Nonfammable Gas	Oxidizer and Nonfammable Gas	Nonflammable Gas	
DOT ID No.	UN 1005	UN 1006	UN 1046	UN 1013	UN 1049	UN 1066	UN 1072	UN 1002	
CAS No.	7664-4-7	7440-37-1	7440-59-7	124-38-9	1333-74-0	7727-37-9	7782-44-7		
Compatible	101101		Noncorrosive; most common structural materials can be used.	Noncorrosive; most common structural materials can be used. If molisture is present, materials must resist carbonic acid	Noncorrosive; most common structural materials can be used	Noncorrosive; most common structural materials can be used.	Noncorrosive; structural materials must be suitable for oxygen service	Noncorrosive; most common structural materials can be used.	

^{*} Strong oxidizer, regarded as non-toxic, but exposure to high concentrations adversely affects pulmonary and central nervous systems. Supports and vigorously accelerates combustion avoid all contact with oil, grease, or other combustible or flammable materials.

^{**} At high pressure can accelerate the burning of combustible and flammable materials